

GATE XL (Biochemistry)

Regulation and Inhibition of Enzymes

1. Multiple Choice Question (MCQ)

Which of the following is an example of **allosteric regulation** of an enzyme?

- A) Competitive inhibition of hexokinase by glucose-6-phosphate
- B) Activation of pyruvate dehydrogenase by ATP
- C) Feedback inhibition of aspartate transcarbamoylase by CTP
- D) Non-competitive inhibition of succinate dehydrogenase by malonate

Answer: C) Feedback inhibition of aspartate transcarbamoylase by CTP

2. Numerical Answer Type (NAT)

An enzyme follows Michaelis-Menten kinetics and has a **K_m of 5 mM** and a **V_{max} of 200 μmol/min**. In the presence of a **competitive inhibitor**, the apparent K_m increases to **10 mM**, while V_{max} remains unchanged. Determine the inhibitor's **inhibition constant (K_i)** if the inhibitor concentration is **5 mM**.

(Use the equation: $K_i = [I] / ((K_m' / K_m) - 1)$)

Answer: $K_i = 5 \text{ mM} / ((10 / 5) - 1) = 5 \text{ mM} / 1 = 5 \text{ mM}$

3. Multiple Choice Question (MCQ)

Which type of enzyme inhibition **cannot** be overcome by increasing the substrate concentration?

- A) Competitive inhibition
- B) Non-competitive inhibition
- C) Uncompetitive inhibition
- D) Both B and C

Answer: D) Both B and C

4. Multiple Select Question (MSQ)

Which of the following statements are **true** regarding enzyme regulation?

- A) Allosteric enzymes do not follow Michaelis-Menten kinetics
- B) Phosphorylation can either activate or inhibit enzyme activity
- C) In competitive inhibition, V_{max} decreases
- D) Feedback inhibition is a form of metabolic regulation

Answer: A, B, and D

5. Numerical Answer Type (NAT)

A particular enzyme has a turnover number (**k_{cat}**) of **500 s⁻¹** and a total enzyme concentration (**[E]_t**) of **0.02 mM**. Calculate the **maximum velocity (V_{max})** of the enzyme reaction in **μmol/min**.
(Use the formula: $V_{max} = k_{cat} \times [E]_t$)

Answer:

$$V_{max} = (500 \text{ s}^{-1}) \times (0.02 \text{ mM})$$

$$= 10 \text{ mM/s}$$

$$= 600 \text{ μmol/min}$$

